## Carr Head Primary School - Knowledge Organiser

**Properties & Changes of Materials** 

Year 5

Autumn

## **Key Knowledge**

Some changes are **reversible** (can be changed back) whilst others are **irreversible** (cannot be reversed).

A new material is always formed after an irreversible change

A mixture is created when two or more materials are combined and can be separated using methods such as sieving,

Sometimes when a **solid (solute)** is mixed with a **liquid** (**solvent**) it will dissolve to form a **solution**.

The solid seems to disappear in the solution.

Science

A **soluble** material can dissolve however an **insoluble** material cannot dissolve.

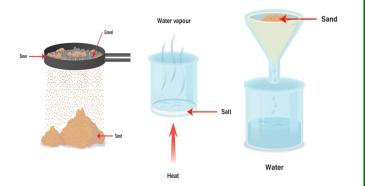
More will dissolve in a hot liquid rather than a cold liquid.

When no more solid can dissolve, the solution becomes satu-

**Sieving** is a mixture of different solid particles can be separated using a sieve.

**Filtering** is where an insoluble solid can be separated from a liquid when passed through a filter. The liquid can pass through the filter whilst the solid particles are trapped in the

**Evaporating** is when salt is mixed with water, it forms a solution. The salt seems to disappear in the water. If the solution is boiled, the solid can be recovered. The water will evaporate



| Know how to          |  |  |
|----------------------|--|--|
| Working Scientically | Compare & group together everyday materials            |  |
|                      | Know some materials will dissolve in liquid to form a  |  |
|                      | solution, and describe how to recover a substance      |  |
|                      | Decide how mixtures might be separated, including      |  |
|                      | filtering, sieving and evaporating                     |  |
|                      | Demonstrate that dissolving, mixing & changes of state |  |
|                      | are reversible changes                                 |  |

| Key Vocabulary                |  |  |  |
|-------------------------------|--|--|--|
| Soluble                       | Solids that will dissolve in a liquid are described as soluble.  |  |  |
| Insoluble                     | Solids that won't dissolve are insoluble.  |  |  |
| Electrical<br>conduc-<br>tors | These materials allow electricity to pass through them e.g. copper inside electrical wires   |  |  |
| Electrical insu-<br>lators    | These materials do not allow electricity to pass through them e.g. the plastic coating on electrical wires.  |  |  |
| Thermal con-<br>ductors       | Materials allow heat to pass through them e.g. a metal saucepan.   |  |  |
| Thermal insula-<br>tors       | Materials do not allow heat to pass through them e.g. a wooden spoon.  |  |  |
| Change of state               | When a material changes from being a solid to a liquid, a liquid to a gas, a gas to a liquid or a liquid to a solid. These are reversible changes  |  |  |
| dissolve                      | When a solid becomes incorporated into a liquid to form a solution   |  |  |
| evaporation                   | When a liquid turns into a gas. When this<br>happens to a solution, the solid is left<br>behind  |  |  |
| filter                        | Any of several types of equipment or devices for removing solids from liquids or gases e.g. cloth, filter paper and sand.  |  |  |
| mixture                       | A substance made by mixing other substances together which can be separated by filtration.   |  |  |
| sieve                         | A tool used for separating solids from liquids or larger pieces of something from smaller pieces   |  |  |
| rusting                       | The corrosion of the metal iron when it comes into contact with water and oxygen, it is an example of an irreversible change.  |  |  |
| Solution                      | A liquid mixture where a solid has dis-<br>solved into a liquid.   |  |  |
| Saturated solu-<br>tion       | A solution that is full of a dissolved substance and cannot dissolve any more.   |  |  |
| Reversible<br>changes         | A change in materials that can be reversed. For example; water can be frozen and change to ice. By heating the ice, this change can be reversed and you can return the ice to its original liquid form of water                        |  |  |
| Irreversible<br>changes       | When a chemical reaction occurs between substances that have been mixed together e.g. bicarbonate of soda and vinegar or burning a material. These changes cannot be undone as another chemical substance has been produced during the |  |  |